

Application of Biogas: Review

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ABSTRACT: The processing of biogas is a well-established technique, mainly in the production of renewable energy and also for the recycling of organic waste. Biogas is the end result of the so-called anaerobic digestion process, a biological controlled method wherein various microorganisms adopt various metabolic processes to decompose organic material. The system has been recognized for centuries and has been commonly used for centuries to domestic households that provide heat and electricity. Today, the biogas industry is increasing increasingly and recent achievements are paving the foundations for developing biogas plants as advanced bio-energy plants. In this case, the idea of a circular economy promoting fertiliser recovery, greenhouse gas emission mitigation and bio-refinery purposes is based on biogas plants. This study outlines the existing state-of-the-art and provides potential prospects for biogas development relevant to the anaerobic digestion process. In addition, a historical retrospective of the biogas industry from the early years of its growth to recent progress offers an outline of the prospects that are opening up for process optimization.

Keywords: Biogas, Bio-energy, Bio Waste, Emissions, Fuel, Renewable energy source, Organic material.

INTRODUCTION

Anaerobic degradation or digestion (AD) is a microbial mediated mechanism wherein organic carbon is transformed to its most oxidised state (CO2) and to its most reduced state by subsequent oxidation and reduction (CH4). A wide variety of microorganisms acting synergistically in the lack of air are catalysing this biological pathway. It is well known that in numerous habitats, including wetlands, rice fields, intestines of animals, marine sediments and manures, AD is responsible for carbon recycling. This procedure is also used widely for the valorisation of organic residues on an industrial scale. In many nations, the disposal of waste and waste water has been a national concern.Generally, biowastes, i.e. sludge, compost, agricultural or industrial organic waste, as well as polluted soils etc. have been used in raw soils as biofertilizers or discarded in landfills, or, in the worse situations, discharged into the atmosphere.

Nevertheless, environmental awareness has enacted stringent regulations banning such behaviours. For example, strict permitting guidelines for the disposal of biodegradable organic matter in landfills have been developed by the European Union. In certain situations, the management of biowastes by AD processes is the best method of transforming agricultural waste into usable items such as electricity (in the form of biogas) and soil conditioning (fertilizer). This basically ensures that after the biowastes have been stabilised, the residual residues can be transferred to the agricultural soils by removing the energy potential, supplying all the required beneficial nutrients and preserving humus and soil structure [2]. Compared to the aerobic process of bacterial biomass, the key benefits of the industrial AD process depend on the creation of a flexible energy carrier and the high degree of organic matter reduction with minor increases. The aim of this analysis is to summarise



current awareness of the process of biogas production and to describe innovative developments that are expected to play a strategic role in the immediate future.

BIOGAS AND ITS UTILIZATION FOR ENERGY PRODUCTION

Depending on the influential feedstock, the applied temperature and reactor configuration, there are distinct ways to define the operating mode of biogas plants. The type of anaerobic digestion reactor is primarily determined by the quality and dry matter composition of the controlling agent to be processed. Complete Suspended Solids (TSS) reactors with flocculent sludge can be used for successful substrates below 500 mg/L. Immobilized granular sludge style reactors such as UASB or EGSB may be used for the elevated TSS concentration in the prominent substrates. Finally, for slurries, like manures, Continuous Stirred Tank Reactors (CSTR) are most widely used with TSS varying from 30 to 70-80 g/L.Specific types of rector configurations were designed for higher dry matter content substrates (> 100 g/L), taking into account the mixing and transport of solid influences. Both dry and wet fermentation, an initial difference may be described. The word "dry fermentation" defines the degradation process, that is characterised by a high content of solids varying from 15 percent to 35 percent (or even higher for solid waste batch garage style reactors), whereas on the opposite, the solid content is up to 10 percent during "wet fermentation" and therefore the liquid content is relatively higher. The original structure design of the plant relies on the preference between these two fermentation processes. It must be remembered that, depending on their chemical composition, the methane yield differs greatly between different substrates[1].

A mono-digestion technique is used by only few biogas plants (i.e. the digester processes only a single feedstock). Due to weak methane potential, elevated concentrations of inhibitors (e.g. phenols, ammonia etc.) or seasonal supply of particular substrates, the majority of biogas plants adopt co-digestion feeding techniques. Various organic residues, typically with dissimilar properties, are processed concurrently in the same anaerobic digester during the co-digestion concept [5]. It is important to outline the advantages of the codigestion process as follows:

- Depending on the chemical composition of the substrates used, the loading of readily biodegradable matter increases.
- Enhance the buffer potential of the powerful mixture to preserve the pH levels for methanogenesis within the range.
- Include a better mix of nutrients, particularly in order to improve the C/N ratio.
- Dilute the inhibitors that prevent the anaerobic digestion process from degrading.
- Leading to higher methane volumetric production.
- Enhances synergistic impacts which contribute to advanced biodegradation.
- Adds to seeking a solution to the stirring or pumping of the digesters, especially when processing solid waste.
- Enhances biogas farm economics.
- Gives excellent stabilisation of sanitation.

Main operational parameters influencing the biogas process

Temperature



In anaerobic reactors that run under mesophilic (30 °C-40 °C, primarily 35 °C- 37 °C) or thermophilic (50 °C-60 °C, mainly 52 °C-55 °C) temperature conditions, the total digestion phase occurs. The choice of operating temperature and its control at controlled temperatures is of crucial importance as these factors have a significant effect on the production of the microbial structure of the digesters. Fluctuations in temperature induce process imbalances associated with volatile fatty acids (VFA) aggregation and concomitant reductions in the production of biogas. Thermophilic conditions are very well established to offer a variety of advantages relative to mesophilic ones, namely:

- Smaller reactor hydraulic retention time (HRT), which normally lasts 15 days in thermophilic conditions and 20-25 days in mesophilic conditions.
- Long-chain fatty acids can be further degraded (LCFA).
- Dependent on the chemical structure of the substrates used, it creates lower concentrations and more qualitative effluent digestate.
- Enhances the process's energy balance and decreases the initial capital expense for expenditure due to the reduced scale of the reactor.
- Achieving improved effluent hygiene.
- Due to faster reaction speeds, higher organic loads could be overcome

This is the primary explanation for thermophilic temperature selection. Any rules, including a minimum guaranteed keeping time at a specific thermophilic temperature, must be met in order to ensure good effluent consistency. Many thermophilic biogas plants thus guide the effluents into a holding tank where the effluents are stored for several hours in order to ensure proper sanitation.

pH and volatile fatty acids

The method of processing biogas occurs at a given narrow pH interval varying from about 6 to 8.5. If the pH of the reactor reaches those thresholds, the operation will deteriorate, resulting in a drastic reduction in the production of methane. Changes in pH values can be associated with other operating conditions, so organic acid deposition (acidification) will usually reduce the pH, whereas higher amounts of ammonia or CO₂ elimination can lead to higher pH values. It must be noted that the reduction in pH due to the aggregation of VFA is also contingent on the substrate used.Few organic compounds have a high buffer capacity, such as cattle manure, and are thus able to keep the system's pH healthy. A decrease in pH can occur only in situations where the VFA concentration is remarkably high above a certain threshold, and often the mechanism is already seriously impaired. The deposition of VFA can thus be seen as a consequence of an already suppressed mechanism and is not known as the real cause[2].

Inhibitors of the process

During AD, there's certain substances which can decrease the yield of biogas if their concentration reaches such limits or can cause fatal degradation of the process in the worst circumstances. Either hazardous agents or intermediate metabolic products are these molecules. In particular, contrasted with bacteria, methanogens are found to be more susceptible to future damage to toxicants. The enhanced concentration of ammonia is one of the most popular AD process inhibitors. In a vast range of organic residues, such as swine or



poultry waste and high protein sludge, ammonia is present. In addition, during protein degradation, ammonia may also be produced, or may derive from other substances, like urea. It is well known that free ammonia (NH₃) and not ammonium ion (NH₄+) are due to the inhibition effect [7]. It is widely stated that a concentration of total ammonia nitrogen between 1.7 and 14 g/L may result in a 50 percent decrease in the production of methane. However, it is impossible to quote the absolute concentration value above which ammonia results in process inhibition, since this is additionally contingent on other considerations, including the source of temperature, pH or inoculum.More precisely, free ammonia is in equilibrium with ammonium ion and relies on the pH value for its concentration. Likewise, the operating temperature is influenced by the equilibrium; elevated temperatures contribute to higher free ammonia concentrations, resulting in phenomena of more extreme toxicity. The inhibition of ammonia also induces the aggregation of VFA that in turn decreases the reactor's pH. As the concentration of free ammonia will be lowered, reducing the pH will partly mitigate the toxicity effect of ammonia. This homeostatic mechanism will, however, sustain the reactor's activity in a relapsed period, called the "inhibited steady state" condition. Many full-scale biogas plants running under inhibited steady-state conditions due to high ammonia loads have been reported to lose up to 30 percent of their potential methane output yield, clearly contributing to significant operational problems and severe economic losses[3].

End-use of biogas in the energy sector

Biogas consists mainly of carbon dioxide (25 percent-50 percent) and methane (50 percent-75 percent), but there are also small amounts of nitrogen, hydrogen, ammonia and hydrogen sulphide (usually less than 1 percent). Biogas is traditionally used for heat output or for combined heat and power generation (CHP). Particularly in developing countries where electricity is scarce and people rely on the use of biomass to fulfil their energy requirements, biogas is used extensively to power cooking stoves and to supply lightning. In these regions, biogas reactors are household-scaled with a standard size of only 2-10 m3 that does not allow CHP or purification processes to be accommodated. On the contrary, the produced gas is burned in a CHP unit in farm-scale or consolidated biogas plants, and is converted to approximately 35 percent-40 percent electrical energy, 45 percent-50 percent heat, and 15 percent energy losses depending on the performance of the engine [9]. It should be remembered that, in order to prevent any degradation or deterioration of the combustion engines, the impurities present in biogas and, in particular, hydrogen sulphide must be eliminated. In addition, organic bound minerals and salts are emitted and stored in the reactor's effluent stream as a result of carbon transformation, which can be further used as soil conditioners (i.e. biofertilisers). More focus is now paid to the extension of the use of biogas as a transport fuel or as a replacement for natural gas, as will be further addressed. In order to do so, it is important to eliminate the impurities present in biogas and, in particular, CO₂. This has resulted in improved processes for cleaning and purification, leading to greater business demand for the biogas sector.

CONCLUSIONS

The method of processing biogas is a proven energy-generation technique. Latest events, nevertheless, have opened new horizons for biogas exploitation, widening its future applications. Given the rapid growth of the biogas industry, it is envisaged that more sophisticated process monitoring and control would ensure greater usage of the treated



biomass. A better knowledge of microbial insight can play a particularly significant role in customising the biogas process and deciphering the "black box" anaerobic digestion. Eventually, biogas plants are projected to be innovative bioenergy factories for cleaner and more stable activities in the future.

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